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# Class 10 - ICSE CHEMISTRY

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ICSE Class 10 Chemistry | Organic Chemistry | DPP Questions

# Organic Chemistry Daily Practice Problems

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# ICSE Class 10 Chemistry | Organic Chemistry | DPP Questions

# **Question 1:**

What is organic chemistry? [Level: Easy]

# **Question 2:**

What is Isomerism? [Level: Easy]

# **Question 3:**

Define Hydrocarbons? [Level: Moderate]

#### **Question 4:**

What is a homologous series? [Level: Difficult]

#### **Question 5:**

Write the structural formula for the given compounds.

- 1) 4-Methyl-2-butyne
- 2) 2-Butene [Level: Moderate]

#### **Question 6:**

Define addition reactions with an example? [Level: Moderate]

# **Question 7:**

Write the balanced chemical equations for the following reactions:

- 1) When ethane reacts with chlorine.
- 2) When ethane gas is treated with HCl.
- 3) When ethane is burnt in excess of oxygen. [Level: Easy]

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# **Question 8:**

Why is methane (CH4) called a greenhouse gas? [Level: Easy]

# **Question 9:**

What are the harmful effects of ethanol on the human body? [Level:

Moderate]

# Question 10:

What is glacial acetic acid? [Level: Moderate]

# Question 11:

What are the physical and chemical properties of alcohol? [Level: Moderate]

# Question 12:

What are the sources of alkanes alkynes and alcohols?

# Question 13:

List out the uses of acetic acid? [Level: Moderate]

# Question 14:

What are carboxylic acids? Give the general formula of carboxylic acids. [Level:

Easy]

# Question 15:

Explain in detail how methane  $(CH_4)$  and ethane  $(C_2H_6)$  are prepared in the

laboratory? [Level: Moderate]

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# **Question 16:**

What are the uses of ethene? [Level: Moderate]

# Question 17:

Define substitution reaction [Level: Moderate]

#### Question 18:

Give the structure for the following compounds?

- 1. Prop-1-ene.
- 2. Ethanol.
- 3.3-hexene.
- 4. Vinegar.
- 5. Acetone. [Level: Easy]

# Question 19:

What is pyrolysis or cracking? [Level: Difficult]

# Question 20:

What happens when ethylene  $(C_2H_4)$  is passed through the alkaline KMnO<sub>4</sub>

solution? [Level: Moderate]

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